Absorption correction: $\theta_{max} = 26.3^{\circ}$ empirical via ψ scans $h = 0 \rightarrow 9$ (Fair, 1990) $k = -17 \rightarrow 0$ $T_{min} = 0.969, T_{max} = 0.979$ $l = -11 \rightarrow 11$ 2312 measured reflections3 standard reflections2062 independent reflectionsfrequency: 120 min

Refinement

Refinement on F R = 0.048 wR = 0.060 S = 1.681715 reflections 170 parameters H atoms: see below $w = 1/[\sigma(F)^2 + (0.02F)^2 + 1.0]$ $(\Delta/\sigma)_{\text{max}} = 0.01$ $\Delta\rho_{\text{max}} = 0.22 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{\text{min}} = -0.25 \text{ e} \text{ Å}^{-3}$ Extinction correction: none Scattering factors from International Tables for X-ray

intensity decay: 1%

Crystallography (Vol. IV)

Table 1. Selected geometric parameters (Å, °)

C14—C5	1.411 (2)	C3—C4	1.362 (3)
C14—C13	1.441 (2)	C9-C10	1.359 (3)
C14C1	1.407 (2)	C1—C2	1.370 (3)
C5-N6	1.386(2)	C2-C3	1.391 (3)
C4—C5	1.408 (3)	CI-N8	1.467 (2)
N6—N7	1.284 (3)	C11—C12	1.377 (3)
C8-C13	1.412(2)	N801	1.218 (2)
C8—C9	1.408 (3)	N8	1.211 (2)
C8—N7	1.388 (2)	C10-C11	1.395 (3)
C13C12	1.403 (3)		
C5-C14-C13	116.5(1)	C8-C9-C10	120.0 (2)
C5-C14-C1	115.2 (2)	N6N7C8	120.6(1)
C13-C14-C1	128.4 (1)	C1—C2—C3	119.6 (2)
C14-C5-N6	123.4 (2)	C14—C1—C2	123.5 (2)
C14-C5-C4	121.4 (1)	C14C1N8	120.8 (2)
N6C5C4	115.2 (2)	C2C1N8	115.6(1)
C5-N6-N7	120.0(2)	C4—C3—C2	119.7 (2)
C13-C8C9	120.7 (2)	C13-C12-C11	121.0(2)
C13-C8-N7	123.5 (2)	C1-N8-O1	117.6(1)
C9-C8-N7	115.9 (2)	C1N8O2	118.1 (2)
C14-C13-C8	115.6(1)	O1-N8-O2	124.2 (2)
C14—C13—C12	126.9 (2)	C9-C10-C11	120.3 (2)
C8-C13-C12	117.5 (2)	C12-C11-C10	120.4 (2)
C5-C4-C3	120.5 (2)		
C13-C14-C5-N6	5.2 (2)	C14—C5—C4—C3	-0.8(3)
C13-C14-C5-C4	-176.2 (2)	C9-C8-C13-C12	4.0 (3)
C1-C14-C5-N6	-175.1 (2)	C14-C1-N8-O1	-65.7 (2)
C1-C14-C5-C4	3.6 (2)	C14-C1-N8-O2	116.1 (2)
C5-C14-C13-C8	-7.3 (2)	C2-C1-N8-O1	110.4 (2)
C13C14C1N8	-8.1 (3)	C2-C1-N8-O2	-67.8(2)

The structure was solved by direct methods. Most of the Hatom positions were located by difference synthesis and refined isotropically; the remaining ones were calculated geometrically and a riding model was used during the refinement process.

Data collection: *MolEN* (Fair, 1990). Cell refinement: *MolEN*. Data reduction: *MolEN*. Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1990). Program(s) used to refine structure: *MolEN*. Molecular graphics: *ORTEPII* (Johnson, 1976). Software used to prepare material for publication: *MolEN*.

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Homolycorine hydrochloride dihydrate[†]

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Abstract

The chirality of homolycorine has been determined by X-ray crystallographic analysis of the hydrochloride dihydrate, $C_{18}H_{22}NO_4^+$ ·Cl⁻·2H₂O. Homolycorine is an

[†] Alternative name: 9,10-dimethoxy-1-methyllycorenan-7-one hydrochloride dihydrate.

Amaryllidaceae alkaloid, which was isolated from Narcissus confusus pugsley. The structure is compared with those reported for other alkaloids of the homolycorine/lycorine series.

Comment

Homolycorine, which is widely distributed in the Narcissus L. genus (Bastida et al., 1997), is one of the principal constituents of Narcissus confusus (Bastida et al., 1987). This plant, belonging to the pseudonarcissi DC sect., is an endemic Amaryllidaceae species from the centre of the Iberian Peninsula and was collected in Béjar (Salamanca) during the flowering period. The genus Narcissus belongs to the tribe Narcisseae, which is one of nine tribes in this family, and has a typical West-Mediterranean distribution, being extremely variable in Spain and Portugal where most species occur.

According to the literature, homolycorine has been described as a cytotoxic agent against fibroblastic LMTK cells and as an inductor of delayed hypersensitivity in animals (Weniger *et al.*, 1995; Gude *et al.*, 1988).

Little is reported on the chirality of alkaloids of the homolycorine/lycorenine series (Clardy *et al.*, 1972; Gopalakrishna *et al.*, 1978; Via *et al.*, 1989; Latvala *et al.*, 1995), thus the structural chirality hitherto has been reported with reservation. In order to determine it, we have analyzed the HCl salt of homolycorine, (I).



The Flack (1983) parameter for the given coordinates indicates that homolycorine hydrochloride possesses identical chirality to that which has been previously proposed for homolycorine (Bastida *et al.*, 1987; Wagner *et al.*, 1996; Latvala *et al.*, 1995) and opposite to that proposed in 17-epihomolycorine (Gopalakrishna *et al.*, 1978) or in eugenine (Via *et al.*, 1989). Recently, by single crystal X-ray analysis, we have found that O-methyllycorenine hydrochloride, obtained from O-methyllycorenine isolated from Narcissus bujei, possesses the inverse chirality to that described here (Labraña *et al.*, 1999).

The pyran ring shows a skew-boat form. The doublebond character of C3—C4 causes the non-aromatic C₆ ring to have a skew-planar form with C1 and C10b atoms out of the plane defined by the other four atoms. The indoline ring shows an envelope form with N out of the plane defined by the remaining four atoms. This conformation is similar to those observed in other homolycorine compounds. The protonation of the homolycorine molecule is at nitrogen.

Although not all the H atoms were located, hydrogenbonding interactions are indicated by short intermolecular contacts. These involve N—H···O(water) [N···O5 2.701 (3) Å], water···water [O5···O6ⁱ 2.717 (4) Å; symmetry code: (i) 1 + x, y, z - 1], water···Cl⁻ [O5···Clⁱⁱ 3.108 (3); O6···Cl 3.139 (3); O6···Clⁱⁱⁱ 3.196 (3) Å; symmetry codes: (ii) 2 - x, $\frac{1}{2} + y$, 1 - z; (iii) 1 - x, $y - \frac{1}{2}$, 2 - z], and produce a sheet polymeric network parallel to the (101) plane.



Fig. 1. Molecular structure of the homolycorine moiety showing 50% probability displacement ellipsoids. H atoms are shown as circles of an arbitrary radius.

Experimental

Homolycorine was isolated from *Narcissus confusus*. The physical data of the isolated compound (m.p., $[\alpha]_D$, UV, IR, EIMS, ¹H NMR, ¹³C NMR and CD) were in accordance with those reported previously (Jeffs *et al.*, 1985; Bastida *et al.*, 1987; Wagner *et al.*, 1996). Crystals for X-ray analysis were obtained by evaporation of water from an aqueous solution containing HCl.

Crystal data

 $C_{18}H_{22}NO_4^+ \cdot Cl^- \cdot 2H_2O$ Mo $K\alpha$ radiation $\lambda = 0.71069 \text{ Å}$ $M_r = 387.85$ Monoclinic Cell parameters from 25 reflections $P2_1$ a = 11.173(4) Å $\theta = 12 - 21^{\circ}$ $\mu = 0.229 \text{ mm}^{-1}$ b = 7.324(5) Å c = 12.024 (8) Å T = 293(2) K $\beta = 97.82 (4)^{\circ}$ Prism $V = 974.8 (10) \text{ Å}^3$ $0.4\,\times\,0.3\,\times\,0.3$ mm Z = 2Colourless $D_x = 1.321 \text{ Mg m}^{-3}$ D_m not measured

Data collection

Enraf-Nonius CAD-4 diffractometer ω -2 θ scans Absorption correction: none 2326 measured reflections 2214 independent reflections 1774 reflections with $l > 2\sigma(l)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.033$ $wR(F^2) = 0.057$ S = 0.8752214 reflections 281 parameters H atoms treated by a mixture of independent and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0202P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $R_{int} = 0.023$ $\theta_{max} = 29.96^{\circ}$ $h = -15 \rightarrow 15$ $k = 0 \rightarrow 10$ $l = 0 \rightarrow 16$ 3 standard reflections frequency: 120 min

intensity decay: none

$$\begin{split} &\Delta\rho_{\text{max}}=0.175 \text{ e } \text{\AA}^{-3} \\ &\Delta\rho_{\text{min}}=-0.105 \text{ e } \text{\AA}^{-3} \\ &\text{Extinction correction:} \\ &SHELXL97 \text{ (Sheldrick,} \\ &1997\text{)} \\ &\text{Extinction coefficient:} \\ &0.007 \text{ (3)} \\ &\text{Scattering factors from} \\ &International Tables for \\ &Crystallography \text{ (Vol. C)} \\ &\text{Absolute structure: Flack} \\ &(1983) \\ &\text{Flack parameter}=0.06 \text{ (6)} \end{split}$$

Table 1. Selected torsion angles (°)

C6—O1—C1—C10b	51.9 (2)	C2C1C10bC4a	-57.2 (2)
C10b—C1—C2—C3	34.2 (3)	C10-C10a-C10bC1	141.6 (2)
C11—C4—C4a—N	24.2 (2)	C6a-C10a-C10bC1	36.3 (2)
C1-01-C6-02	167.5 (2)	C4a—C4—C11—C12	~ 3.1 (3)
C1 - O1C6C6a	13.6 (3)	C4C11C12N	-19.3 (3)
O1C1C10bC4a	62.0 (2)		

The positions of 13 H atoms were computed. 11 H atoms were located from a difference map and the remaining two (on water molecules) were not located. The computed H atoms were refined isotropically using a riding model and an overall $U_{\rm iso}$, while the remaining H atoms were refined freely.

Data collection: CAD-4/PC (Kretschmar, 1996). Cell refinement: CAD-4/PC. Data reduction: CFEO (Solans, 1978). Program(s) used to solve structure: SHELXS-97 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL97 (Sheldrick, 1997). Molecular graphics: ORTEP (Brueggemann & Schmid, 1990). Software used to prepare material for publication: PLA-TON (Spek, 1990).

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(*3R*)-4,4-Dimethyl-2-oxotetrahydrofuran-3-yl (*2S*)-7-methoxy-2,3-dihydro-1,4benzodioxin-2-carboxylate

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Abstract

The structure of the title compound, $C_{16}H_{18}O_7$, a key intermediate in the synthesis of antagonist adrenergic agents, is reported. The 1,4-benzodioxin ring shows a half-chair form with a Csp^3 atom out of the plane

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